Acta Cryst. (2004). E60, m717–m719 DOI: 10.1107/S1600536804010219 Lo and Ng • [Sn₂BrCl(C₆H₅)₄(C₇H₇)₂(C₆H₅Cl)₄(C₂H₄O₂P₂)] m717

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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.009 Å Disorder in main residue R factor = 0.054 wR factor = 0.114 Data-to-parameter ratio = 18.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title complex, $[Sn_2BrCl(C_7H_7)_2(C_6H_4Cl)_4-(C_{26}H_{24}O_2P_2)]$, the 1,2-bis(diphenylphosphoryl)ethane ligand coordinates to a bromobis(*p*-chlorophenyl)(*p*-tolyl)tin molecule and also to a chlorobis(*p*-chlorophenyl)(*p*-tolyl)tin molecule across a center of inversion, involving disorder of Cl and Br. The Sn atoms are five-coordinate in a *trans*- C_3SnOX geometry.

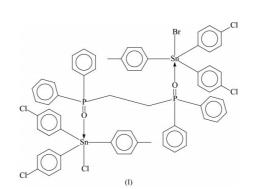
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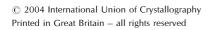
Triphenylphosphine oxide forms a large number of adducts with triorganotin Lewis acceptors (Lo *et al.*, 2001). The bidentate 1,2-bis(diphenylphosphoryl)ethane ligand does not chelate to triphenyltin chloride but instead binds through its double-bond O atoms to two of the Lewis acid molecules (Pelizzi & Pelizzi, 1980). The adduct is centrosymmetric, the center of symmetry being in the middle of the methylene-methylene linkage. A similar centrosymmetric arrangement is found in the mixed bis(4-chlorophenyl)(4-tolyl)bromotin complex (Wei *et al.*, 1990), as well as in the title complex, (I) (Fig. 1). Selected bond distances and angles are given in Table 1. The Sn atom is five-coordinate in a *trans*-C₃SnOX trigonal bipyramidal geometry. The 1/1 adduct of the ligand with butylphenyltin dichloride is also disordered over an inversion site (Ng & Kumar Das, 1996).



Bis(*p*-chlorophenyl)di-*p*-tolyltin was prepared by the reaction of *p*-tolylmagnesium bromide and bis(*p*-chlorophenyl)tin dichloride in THF; the organotin reagent was itself prepared from another reaction between tetra(*p*-chlorophenyltin) (Ng, 1997) and stannic chloride. One of the *p*-tolyl rings was cleaved by using hydrochloric acid to form the mixed bis(*p*-chlorophenyl)(*p*-tolyl)tin chloride. The reagent (0.5 g, 1 mmol) and 1,2-bis(diphenylphosphoryl)ethane (0.4 g, 1 mmol) were each dissolved in ethanol and the two solutions mixed. The compound that separated was collected and recrystallized from chloroform. The molecule of bis(*p*-chlorophenyl)(*p*-tolyl)tin bromide in the crystal structure could have arisen from reaction between

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p-tolylmagnesium bromide and bis(*p*-chlorophenyl)tin dichloride, which could give this compound as a minor product. The compound was probably not removed in the subsequent steps.

 $D_x = 1.555 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation Cell parameters from 25

reflections

 $\begin{array}{l} \theta = 10.0\text{--}13.4^{\circ} \\ \mu = 1.81 \ \mathrm{mm^{-1}} \end{array}$

T = 298 (2) K

 $R_{\rm int} = 0.033$

 $\theta_{\text{max}} = 25.0^{\circ}$ $h = -14 \rightarrow 0$

 $k = 0 \rightarrow 15$ $l = -22 \rightarrow 23$

Block, colorless

 $0.15 \times 0.15 \times 0.10 \ \mathrm{mm}$

3 standard reflections

frequency: 60 min

intensity decay: 3%

 $w = 1/[\sigma^2(F_o^2) + (0.0381P)^2]$

+ 3.4106*P*] where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.48 \ {\rm e} \ {\rm \mathring{A}}^{-3}$

 $\Delta \rho_{\rm min} = -0.60 \text{ e } \text{\AA}^{-3}$

Crystal data

$$\begin{split} & [\mathrm{Sn}_2\mathrm{BrCl}(\mathrm{C}_7\mathrm{H}_7)_2(\mathrm{C}_6\mathrm{H}_4\mathrm{Cl})_4\text{-}\\ & (\mathrm{C}_{26}\mathrm{H}_2\mathrm{O}_2\mathrm{P}_2)]\\ & M_r = 1411.55\\ & \mathrm{Monoclinic}, \ & P_2\ _1/n\\ & a = 12.324\ (3)\ \mathrm{\AA}\\ & b = 12.842\ (3)\ \mathrm{\AA}\\ & c = 19.975\ (4)\ \mathrm{\AA}\\ & \beta = 107.54\ (3)^\circ\\ & V = 3014.4\ (13)\ \mathrm{\AA}^3\\ & Z = 2 \end{split}$$

Data collection

Enraf-Nonius CAD-4 diffractometer ω -2 θ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{min} = 0.744, T_{max} = 0.835$ 5574 measured reflections 5308 independent reflections 3325 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.114$ S = 1.045308 reflections 294 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

Sn1-C1	2.122 (3)	Sn1-O1	2.367 (4)
Sn1-C7	2.125 (3)	Sn1-Cl1′	2.449 (9)
Sn1-C13	2.114 (3)	Sn1-Br1	2.618 (4)
C1-Sn1-C7	120.4 (2)	C7-Sn1-Cl1′	91.4 (6)
C1-Sn1-C13	122.3 (2)	C7-Sn1-Br1	95.4 (2)
C1-Sn1-O1	87.0 (2)	C13-Sn1-O1	83.4 (2)
C1-Sn1-Cl1'	93.7 (7)	C13-Sn1-Cl1'	97.5 (6)
C1-Sn1-Br1	93.8 (3)	C13-Sn1-Br1	93.4 (2)
C7-Sn1-C13	115.8 (2)	O1-Sn1-Cl1'	178.4 (6)
C7-Sn1-O1	87.0 (2)	O1-Sn1-Br1	176.7 (2)

The structure is disordered in terms of the tin-bound Cl and Br atoms, as well as in the chloro/methyl substituent of the tin-bound aryl rings. The occupancy of the pair Br1/Cl1' was initially allowed to refine freely; as this refined to almost 0.5, the occupancy was then fixed at exactly 0.5. Two of the tin-bound aryl rings should be 4-chlorophenyl rings and one the 4-tolyl ring, so that there are two Cl atoms to be shared among three aryl rings. The occupancies of atoms Cl1, Cl2 and Cl3 were refined to 0.85:0.69:0.46. The occupancy of atom Cl1 was then fixed as unity; as the refinement led to occupancies of atoms Cl2 and Cl3 of 0.63:0.37, the occupancies were set at exactly 0.67:0.33. For the Br1/Cl1' pair of disordered atoms, the Sn-Br distance was restrained to 2.65(1) Å and the Sn-Cl distance to 2.45 (1) Å. These values were taken from the literature, the Sn-Brdistance from the structure of the bromotriphenyl adduct with triphenylphosphine oxide (Ng, 1995) and the Sn-Cl distance from the bromotriphenyltin adduct with the same donor ligand (Ng & Kumar Das, 1992). Additionally, the displacement parameters of the two atoms were made equal.

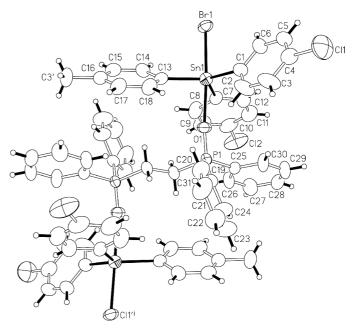


Figure 1

ORTEPII (Johnson, 1976) plot of the title complex (I), showing the atomnumbering scheme and displacement ellipsoids at the 50% probability level. [Symmetry code: (i) 1 - x, 1 - y, 1 - z.]

The Cl-C_{aryl} [1.740 (5) Å] and CH₃-C_{aryl} [1.540 (5) Å] distances were also restrained, more tightly. For each ring, the atoms were restrained to lie in a plane within 0.01 Å. For each methyl group, the distances of the methyl C atom to the ring C atom two bonds apart were restrained to be equal.

All the aromatic rings were refined as rigid hexagons (C–C = 1.39 Å). The H atoms were placed at calculated positions [phenyl C–H = 0.93 Å and $U_{\rm iso} = 1.2U_{\rm eq}(C)$; methylene C–H = 0.97 Å and $U_{\rm iso} = 1.2U_{\rm eq}(C)$; methyl C–H = 0.96 Å and $U_{\rm iso} = 1.5U_{\rm eq}(C)$], and were included in the refinement in the riding-model approximation. The torsion angle of the methyl groups was refined.

Data collection: *CAD-4/PC* (Kretschmar, 1994); cell refinement: *CAD-4 VAX/PC Fortran System* (Enraf–Nonius, 1988); data reduction: *XCAD4* (Harms, 1997) in *WinGX* (Farrugia, 1999); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL*97.

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